

CO₂ storage and injection modelling

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Agenda

- Background and motivation
- Summary on CO₂ storage/trapping mechanisms
- Model review: equations and physics
- Examples:
 1. Solubility trapping
 2. Residual trapping
 3. Poroelasticity
 4. Acidity (pH)

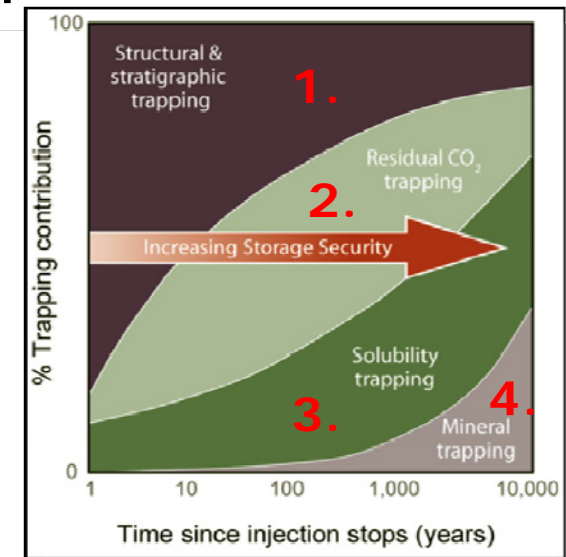
Background and motivation

Two main areas of interest (applications):

1. CO₂ Storage: focus on short term storage and the injection process
2. Shallow gas seepage: gas flow into and out of a well

CO₂ storage/trapping mechanisms

- Storage idea: Inject CO₂ into a **suitable** porous formation below an intact cap rock so that it will not **leak**
- Types of storage mechanisms of CO₂:
 1. **Structural trapping**: Migration blocked by impermeable cap rock layer(s)
 2. **Residual trapping**: CO₂ stabilized by capillary pressure
 3. **Solubility trapping**: In time CO₂ will dissolve in salt pore water (and descend)
 4. **Mineral trapping**: In time CO₂ will chemically react with rock to form carbonates and precipitate



(From: IPCC Special report, 2005)

Modelling challenges

1. How does the CO₂ migrate in/out of the formation, considering perforated/poorly plugged wells, faults, fractures, etc.
 - **Two-phase flow** (maybe even more phases)
2. How does CO₂ injection affect the rock
 - **Poroelasticity** (fracturing)
3. Chemical reactions and borehole integrity
 - **Solubility of CO₂ in water** (Other effects are: chemical reaction with rock formation (eg. calcite), corrosion, erosion, degradation of bore hole due to bad cement plugging)



Model review

Equations and physics



Two-phase flow equations

$$\left. \begin{aligned} \frac{\partial}{\partial t}(\phi \rho_{\alpha} S_{\alpha}) - \nabla \cdot [\rho_{\alpha} \lambda_{\alpha} \mathbf{K}(\nabla p_{\alpha} + \gamma_{\alpha} \nabla z)] &= \rho_{\alpha} q_{\alpha} \\ \sum S_{\alpha} &= 1, \quad p_c = p_n - p_w, \quad S_{e\alpha} = f(p_c) \end{aligned} \right\} \begin{array}{l} \text{General mass} \\ \text{balances and} \\ \text{auxiliary} \\ \text{equations} \end{array}$$



Do some equation manipulation: Eg. fractional flow formulation

(The fractional flow approach treats the multiphase flow problem as a total fluid flow of a single mixed fluid, and then describes the individual phases as fractions of the total flow)

Equations coming up!



Two-phase flow equations

$$\left. \begin{aligned} \frac{\partial}{\partial t} (\phi \rho_\alpha S_\alpha) - \nabla \cdot [\rho_\alpha \lambda_\alpha \mathbf{K} (\nabla p_\alpha + \gamma_\alpha \nabla z)] &= \rho_\alpha q_\alpha \\ \sum S_\alpha &= 1, \quad p_c = p_n - p_w, \quad S_{e\alpha} = f(p_c) \end{aligned} \right\} \begin{array}{l} \text{General mass} \\ \text{balances and} \\ \text{auxiliary} \\ \text{equations} \end{array}$$

⇓

$$\nabla \cdot \mathbf{u} = q_w + q_n - \frac{1}{\rho_w} \frac{\partial \rho_w}{\partial p} \left(\frac{\partial p}{\partial x} u_{w,x} + \frac{\partial p}{\partial y} u_{w,y} - \frac{\partial p}{\partial t} \phi S_w \right) - \frac{1}{\rho_n} \frac{\partial \rho_n}{\partial p} \left(\frac{\partial p}{\partial x} u_{n,x} + \frac{\partial p}{\partial y} u_{n,y} - \frac{\partial p}{\partial t} \phi S_n \right) - \frac{\partial \phi}{\partial t}$$

$$\phi \rho_w \frac{\partial S_w}{\partial t} + \nabla \cdot (\rho_w \mathbf{u}_w) = \rho_w q_w - \phi S_w \frac{\partial \rho_w}{\partial p} \frac{\partial p}{\partial t} - \rho_w S_w \frac{\partial \phi}{\partial t}$$

$$\mathbf{u} = -\mathbf{K} (\lambda \nabla p - (\lambda_w \rho_w + \lambda_n \rho_n) \mathbf{g})$$

$$\mathbf{u}_w = f_w \mathbf{u} + \lambda_n f_w K (\nabla p_c + (\rho_w - \rho_n) \mathbf{g})$$

$$\mathbf{u}_n = f_n \mathbf{u} - \lambda_n f_w K (\nabla p_c + (\rho_w - \rho_n) \mathbf{g})$$



Biot linear poroelasticity equation

Two-ways coupling:

- Flow coupled to geomechanics:

$$\nabla \cdot [\boldsymbol{\sigma}] = -\mathbf{F} \quad \boldsymbol{\sigma} = \mathbf{D}_{el} \boldsymbol{\varepsilon} + \boldsymbol{\sigma}_0 - \alpha_{biot} p \mathbf{I}$$

- Geomechanics coupled to flow:

$$\phi^* = (1 - \varepsilon_v) \phi \quad q_w = q_n = -\alpha_{biot} \frac{\partial \varepsilon_v}{\partial t} \quad K = K_0 \left(\frac{\phi^*}{\phi} \right)^{20}$$

Mass balance equation, CO₂(aq)

$$\frac{\partial(\phi S_w C_{CO_2})}{\partial t} + \nabla \cdot (-D_{CO_2} \phi S_w \nabla C_{CO_2}) = F$$

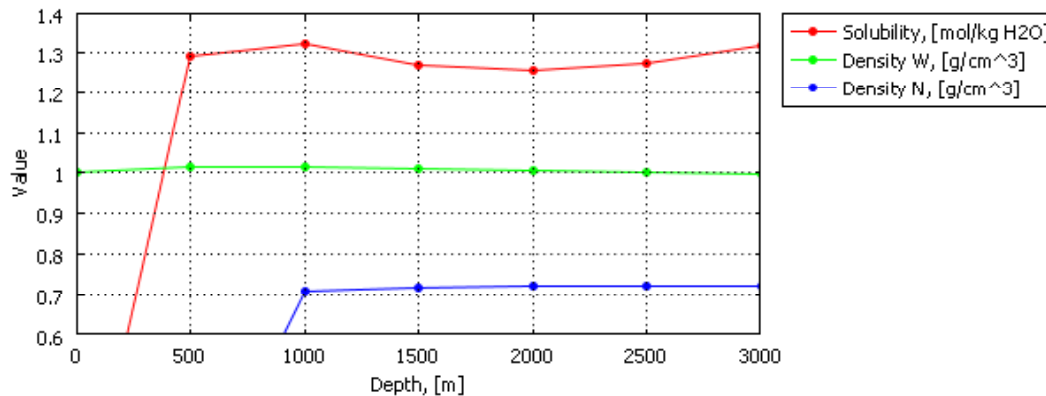
$$\frac{\partial C_{CO_2}}{\partial t} + \nabla \cdot (-D_{CO_2} \phi S_w \nabla C_{CO_2}) = F - \phi C_{CO_2} \frac{\partial S_w}{\partial t} - S_w C_{CO_2} \frac{\partial \phi}{\partial t}$$

$$F = k_s \left(\min(C_{sat,CO_2} S_w, C_{0,CO_2} S_n) - C_{CO_2} \right)$$

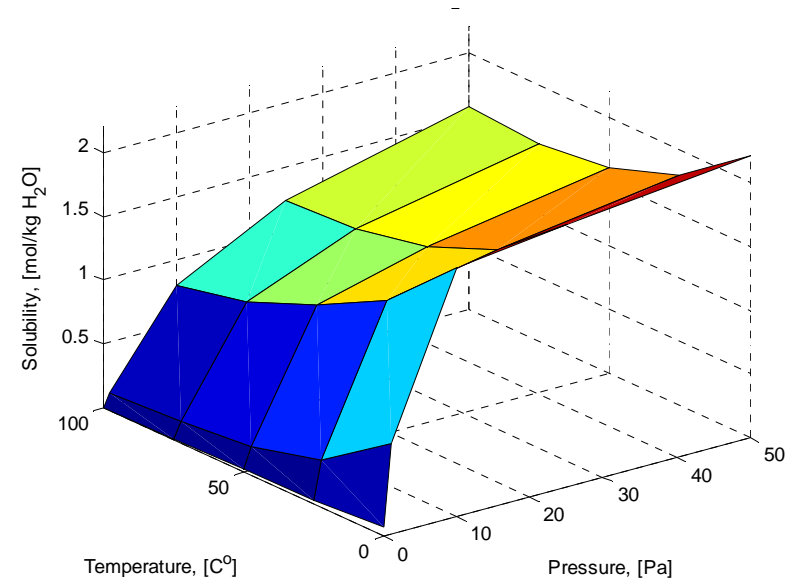
- C_{CO_2} – concentration of dissolved CO₂ in the wetting phase [mol/m³]
 C_{sat,CO_2} – saturation concentration of CO₂ in water; CO₂(aq) [mol/m³]
 C_{0,CO_2} – concentration of pure CO₂ [mol/m³]

CO₂ solubility and densities

Symbol	Value	Unit	Description
ρ_{H2O}	1000	[kg/m ³]	Density of water-phase
ρ_{CO2}	700	[kg/m ³]	Density of CO ₂ -phase
$C_{sat,CO2}$	1300	[mol/m ³]	Saturation concentration of CO ₂ (aq)



(Temperature gradient used: 28 °K/km)



Model definition

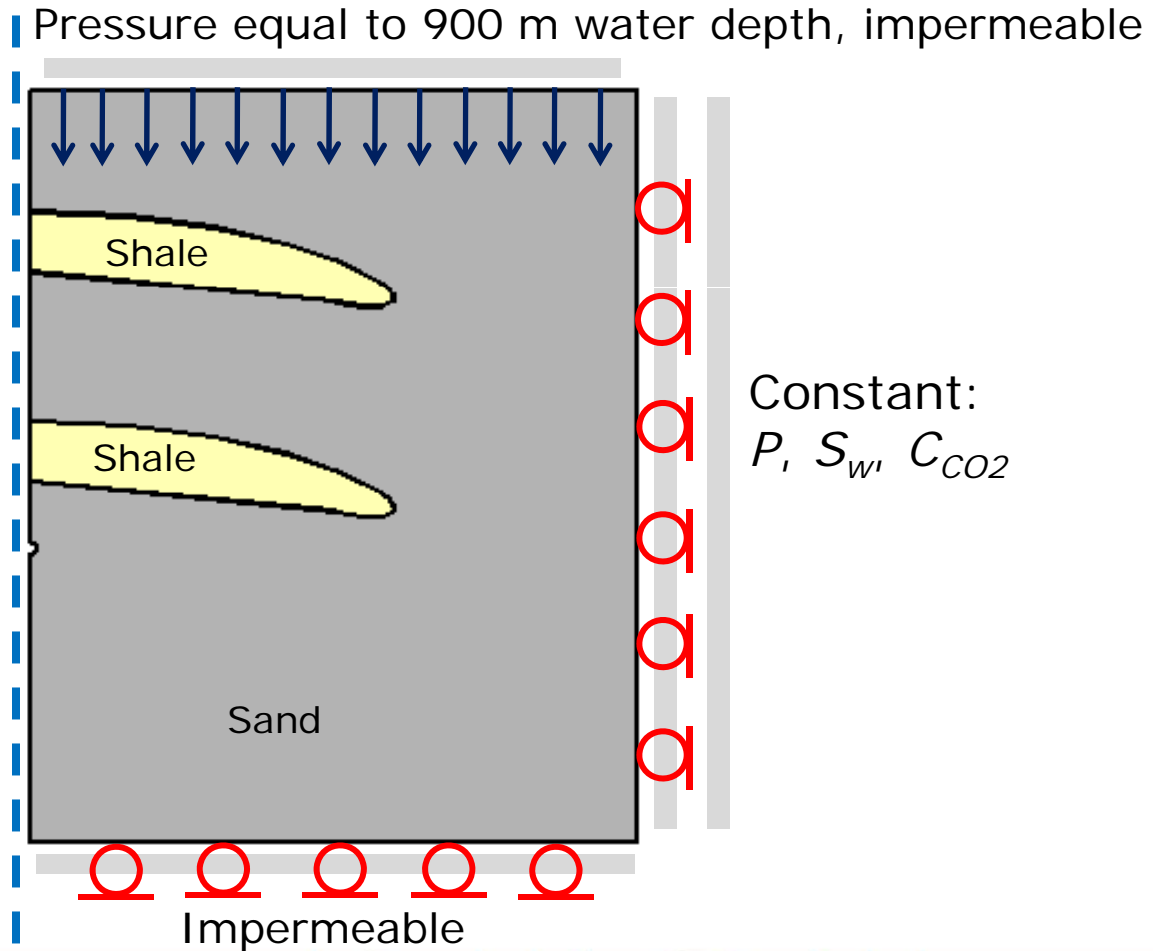


Model (200 x 300 m²), axial symmetry

ϕ [-]	K [D]	P_d [kPa]	E [GPa]	ν [-]
0.4	3	20	2	0.3
0.1	0.01	20	2	0.3

Injection rate:
30 kg/s
(1 Mill. ton/year)

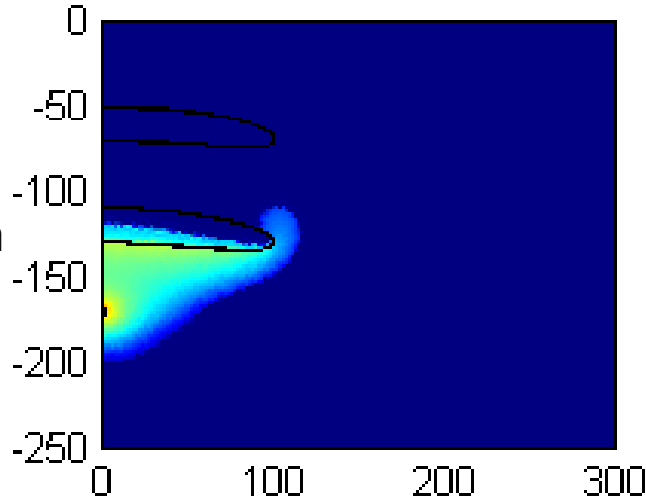
Axial symmetry line



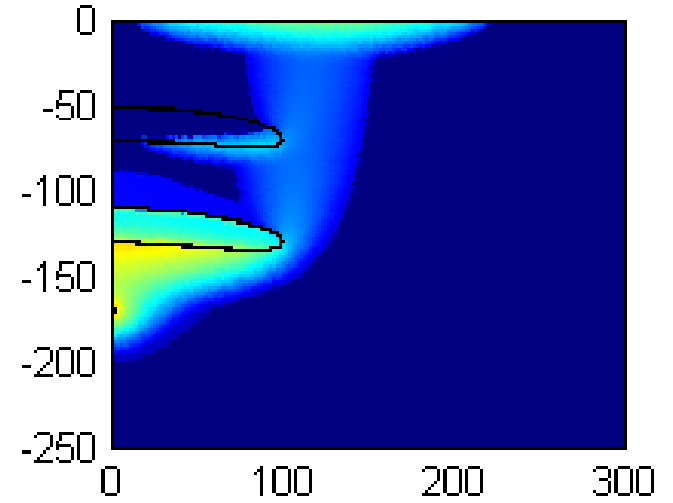
Examples



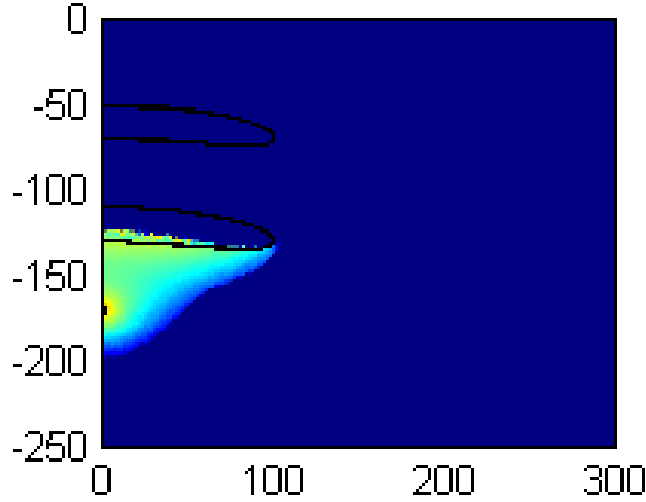
t = 0.1 years, Sen



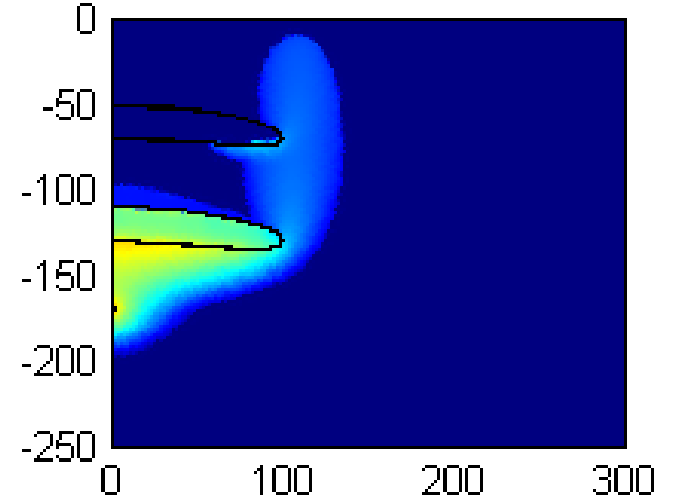
t = 0.5 years, Sen



t = 0.1 years, Sen



t = 0.5 years, Sen



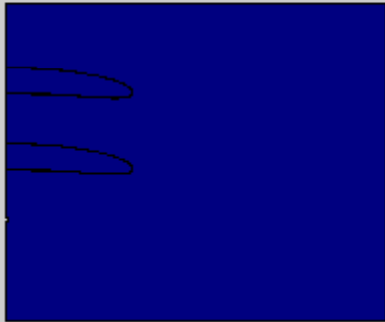
No-dissolution

Dissolution

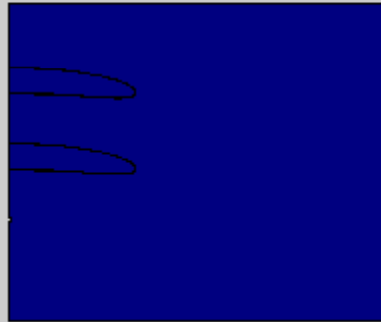
1: Solubility trapping



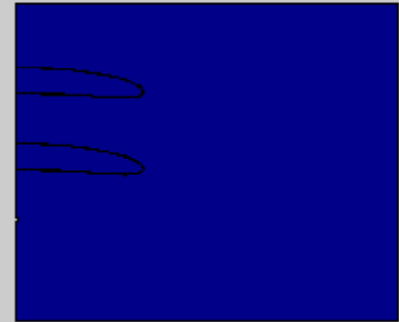
Sen. Time: 0.000 years



Sen. Time: 0.000 years

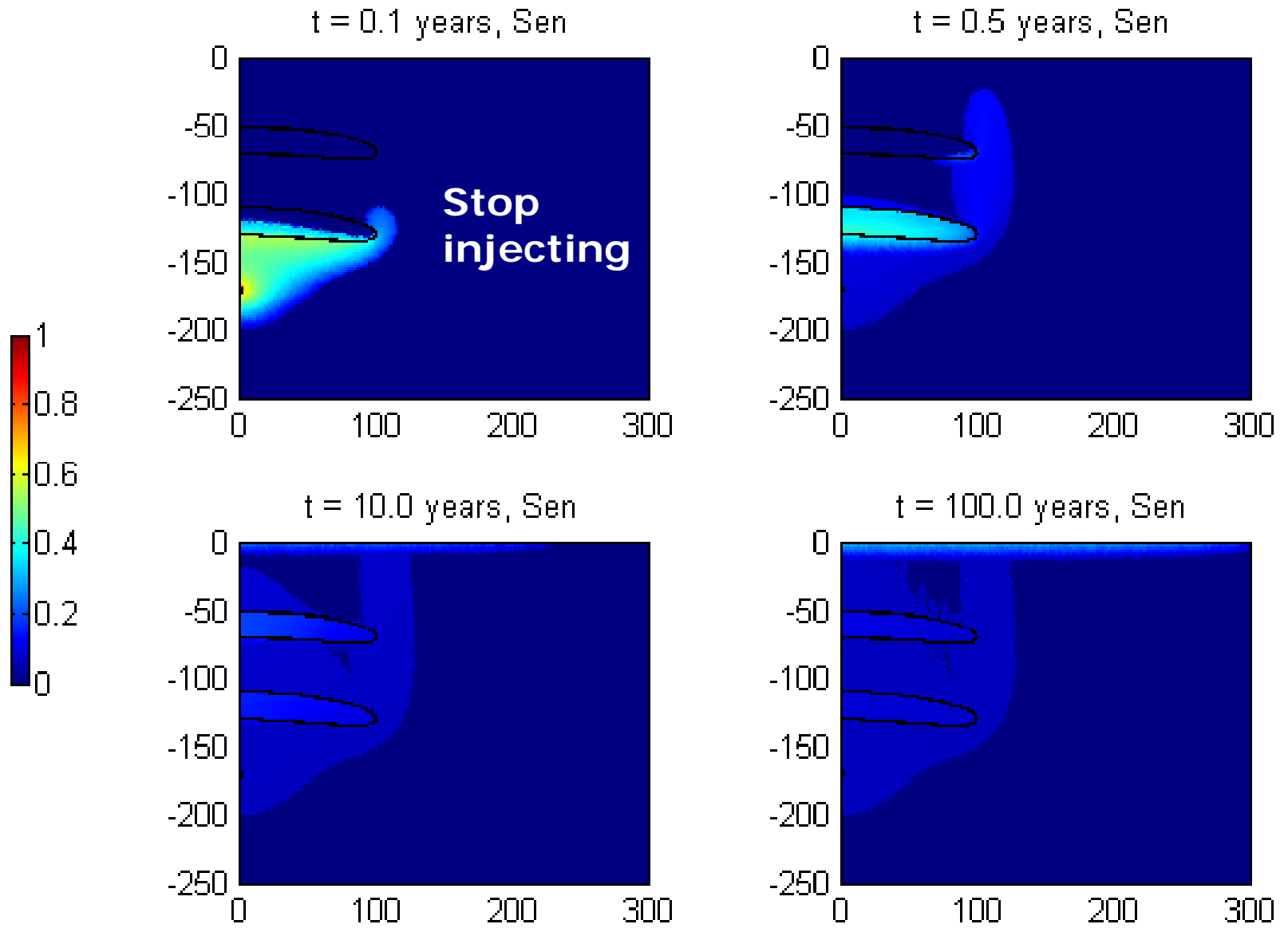


CO₂ Time: 0.000 years



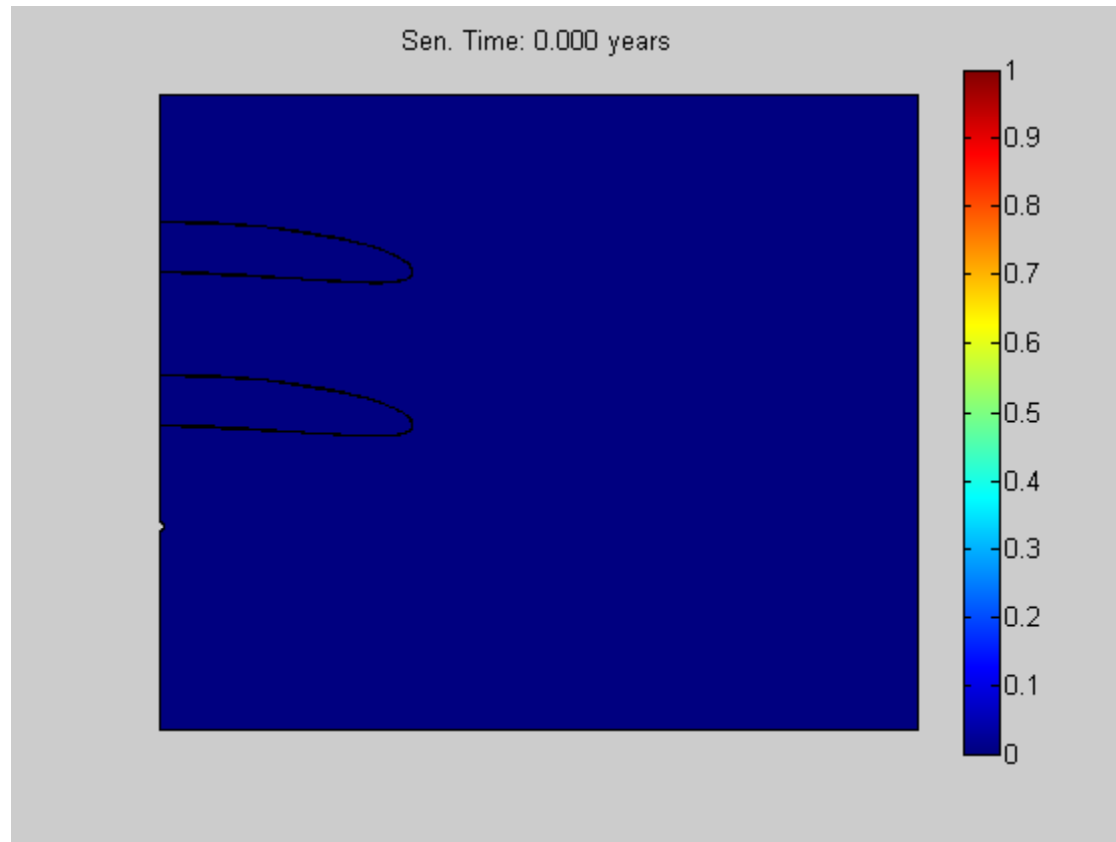
1: Solubility trapping





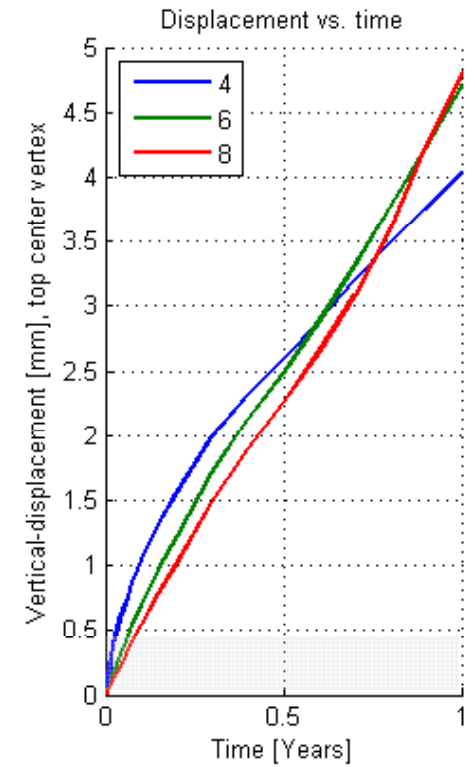
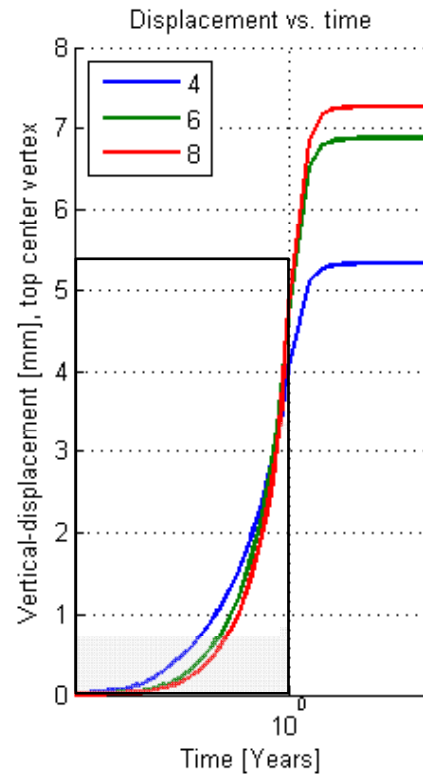
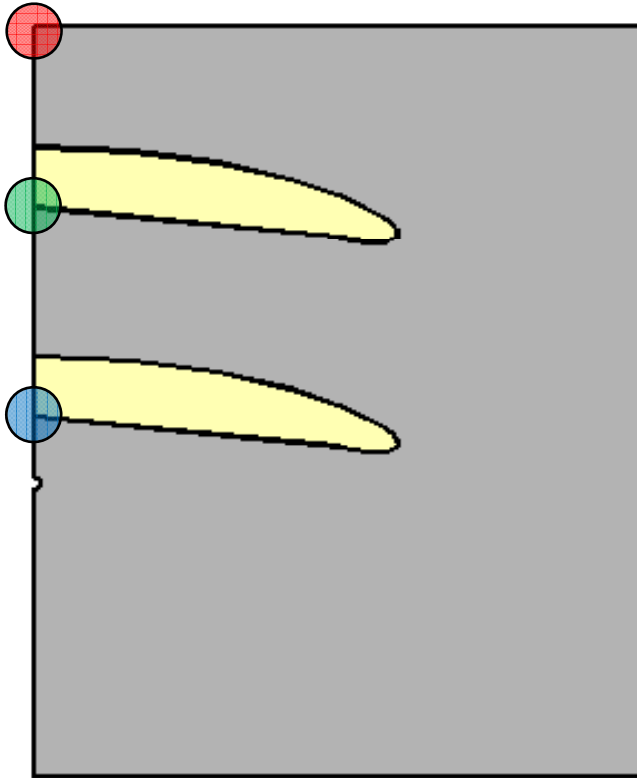
2: Residual trapping ($S_r = 0.05$)





2: Residual trapping ($S_r = 0.05$)

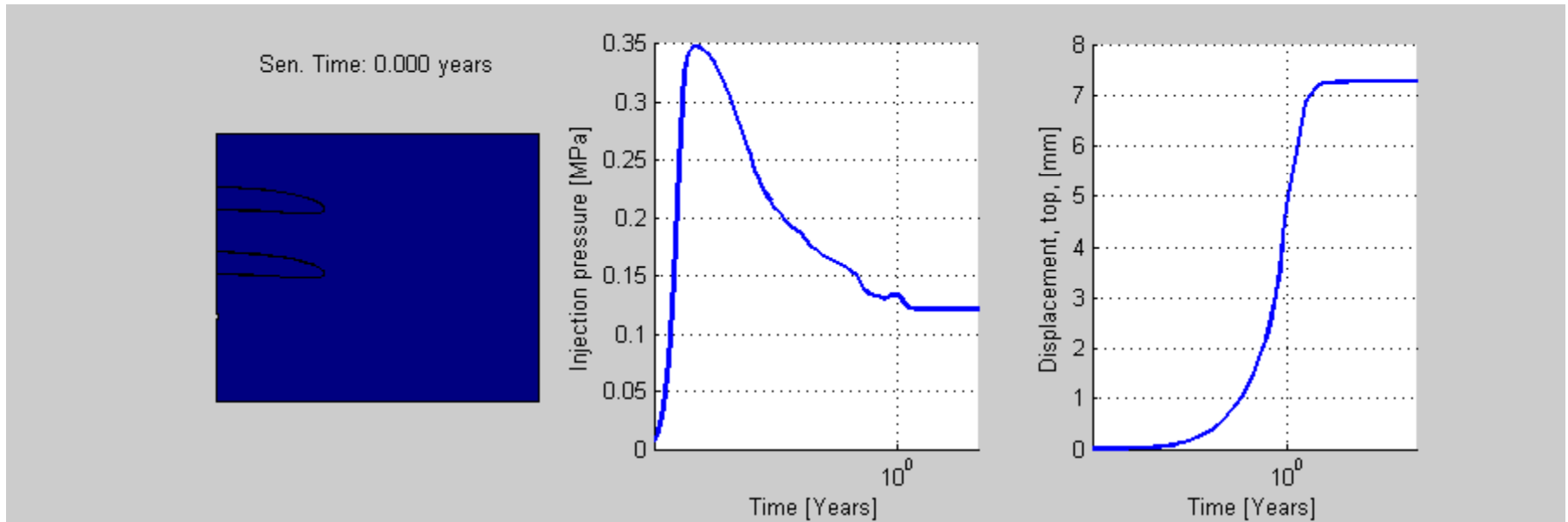
Example 3: Poroelasticity



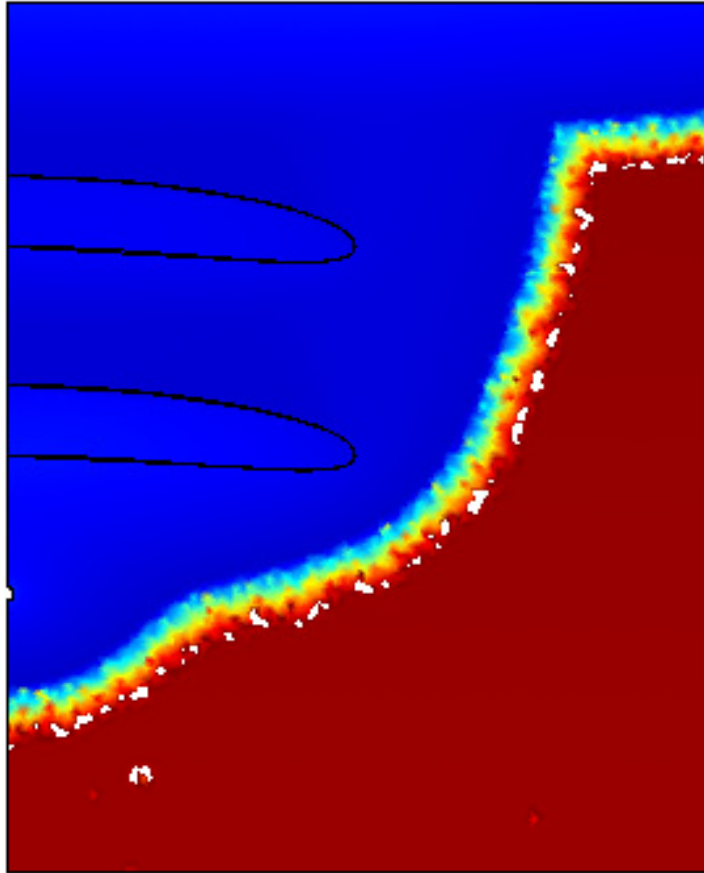
Wetting saturation

Injection pressure

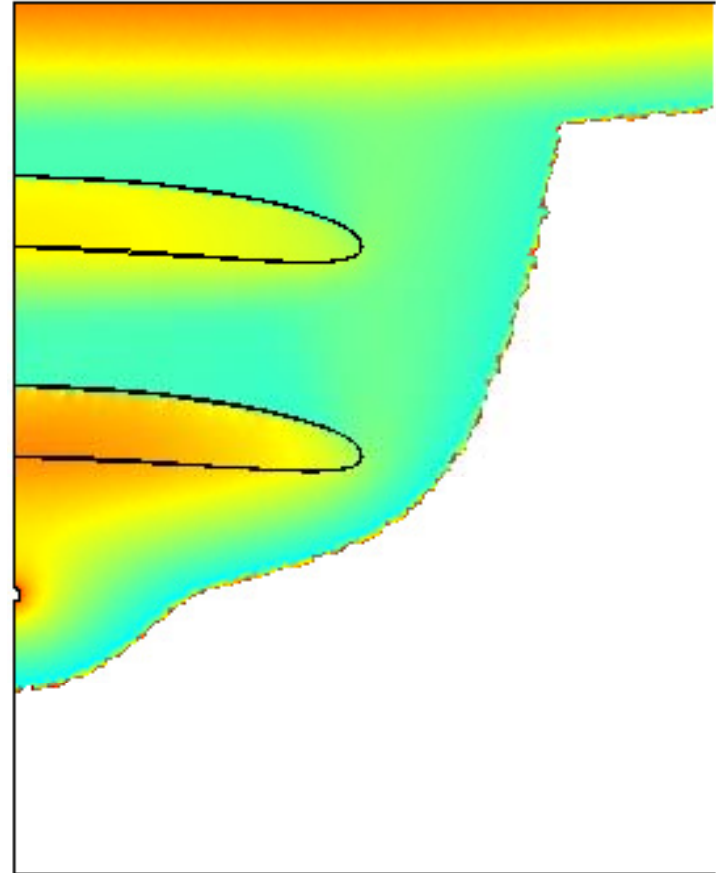
Displacement



t = 10.0 years, pH
Min: 3.2, max: 5.6



t = 10.0 years, pH
Min: 3.0, max: 3.5



pH measurements are good detection methods
However, calcite buffers pH (maybe as much as 2)

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Conclusion

- A robust model that considers many main features related to CO₂ injection has been developed
 - Improvements: More chemical reactions, more realistic geomechanical materials
- Most important short term storage mechanisms:
 - Residual and solubility trapping
- Remaining tasks:
 - Verification (compare with lab-experiment)
 - Simulation of “real-life” cases and risk assessment evaluation
- Thank you for your attention!